

IERT PRAYAGRAJ SYLLABUS

OFFICIAL

11. STRUCTURE OF ENTRANCE EXAMINATION (प्रवेश परीक्षा का ढॉचा)

- (a) Question paper will be of three hours, which will contain multiple choice type objective questions of maximum 300 (Three hundred) marks. There will be an interview of hundred marks of those courses in which interview are part of admission process. प्रवेश परीक्षा में वस्तुनिष्ठ प्रकार का तीन घंटे का एक प्रश्नपत्र होता है जिसके अधिकतम अंक 300 होते हैं जिस कोर्स में साक्षात्कार होता है उसमें साक्षात्कार के लिए 100 अंक अलग होते हैं
- (b) Every question booklet will have 75 multiple choice type objective questions with four options of answer for every question. प्रत्येक पश्नपुस्तिका में 75 वस्तुनिष्ठ (ओब्जेक्टिव) प्रकार के प्रश्न होते हैं जिसमें प्रत्येक प्रश्न के चार उत्तर विकल्प के रूप में होते हैं।
- (c) Candidate has to attempt all the questions. For each correct answer four marks will be awarded whereas one mark will be deducted for each wrong answer. प्रत्येक सही उत्तर के लिए चार अंक दिये जाते हैं तथा प्रत्येक गलत उत्तर के लिए एक अंक काट लिया जाता है। सभी प्रश्न करना अनिवार्य है।
- (d) Admission process is completed strictly according to the merit in the auditorium of the institute at the time of counseling on the scheduled time and date. If a candidate does not report on time at the call of counselor on the date of counseling he/she will be considered absent/not interested and his/her seat of admission will be given to the candidate who stands next to hem/her in merit. प्रवेश मेरिट लिस्ट के अनुसार संस्था के सभागार में काउन्सिलिंग के लिए निर्धारित दिनांक एवं समय पर मेरिट से किया जाता है। यदि बुलाने के समय पर अभ्यर्थी उपस्थित नहीं है तब मेरिट के अगले अभ्यर्थी को बुलाकर प्रवेश प्रक्रिया पूरी करा ली जाती है।
- (e) In any particular day pre decided fixed number of required admission is taken and then counseling is closed. **Bharatgeek** जिस दिन जितने प्रवेश की आवश्यकता होती है, उतने प्रवेश पूरा होने पर उस दिन की काउन्सिलिंग वहीं पर बन्द कर दी जाती है।
- (f) **It is must for the candidate to be present at the place in time as mentioned in his/her offer letter for counseling obtained alongwith their entire original certificates, three photocopies of them, Bank Draft of the fees, twelve copies/ number of the same photographs as affixed in the application form.** अभ्यर्थी को सचेत किया जाता है कि काउन्सिलिंग हेतु उनके द्वारा प्राप्त किये गये बुलावा पत्र में निर्धारित स्थान एवं समय पर स्वयं अपने सभी मूल प्रमाण पत्र (तीन प्रति प्रत्येक की छायाप्रति सहित), फीस का ड्राफ्ट, 12 अदद स्वयं के फोटोग्राफ जैसा कि आवेदन पत्र पर लगाया गया था, उपस्थित हों।

Official Source: IERT

✓ **UPDATED सिलेबस यहाँ देखें**



IERT PRAYAGRAJ SYLLABUS



MATHEMATICS

16. SYLLABUS FOR IERT ENTRANCE EXAMINATION-2025

SYLLABUS FOR THREE YEAR (6 SEMSTER) ENGINEERING DIPLOMA ENTRANCE EXAMINATION-2025

MATHEMATICS

Bharatgeek

A. ARITHMETIC:

1. Simple & Compound interest.
2. Work-time and speed-time.
3. Bank deposits and payments in installments, Taxation.
4. Percentage, Profit, Loss & Discount.
5. Ratio, Proportion & Mixture.

B. STATISTICS:

1. Importance of statistics and its usefulness.
2. Classification of data frequency distribution, cumulative frequency. Graphical representation of statistical data, bar chart, histogram, frequency polygon, frequency curves, cumulative frequency curve and graphical representation, Pie diagram.
3. Measures of Central Tendency- Mean, Median & Mode.
4. Measures of Dispersion – Range, Mean Deviation, Standard Deviation & Variance.

C. ALGEBRA:

1. Number System-Natural Numbers, Whole Numbers, Integers, Even Numbers, Odd Numbers, Rational Numbers, Irrational Numbers, Real Numbers, Prime Numbers, Twin Prime Numbers, Composite Numbers and their characteristics.
2. Laws of Indices, Surds & Simplification.
3. Square Roots & Cube Roots.
4. Logarithms-definition, laws and their application, to find the characteristics & mantissa.
5. Factors-expressions in the form of difference of two squares, trinomial expression, factorization by completing the square, expressions in the form of sum and difference of two cubes, remainder theorem and expression of the type $a^3+b^3+c^3-3abc$.
6. L.C.M. & H.C.M by factorization.
7. Linear & Quadratic Equations.
8. Solution of simultaneous equations (linear equations with three variables) and problems based on them.
9. Set Theory – Sets, Representation of Sets, Types of Sets, Subsets, Universal set, Complement of Set, Union of Sets, Interaction of Sets, Disjoint Set, Difference of Sets, Laws of Algebra of Sets & Simple Problems with the help of Venn diagram.
10. Mappings - Into Mapping, Onto Mapping, One-One Mapping and Many-One Mapping.

D. TRIGONOMETRY:

18

1. Circular measures (relation between radian & degree), definition of trigonometric ratios and their relations.
2. Trigonometrical ratios of angles of 0° , 30° , 40° , 60° , 90° and 180°
3. Trigonometrical ratios of angles of $90^\circ \pm \theta$, $180^\circ \pm \theta$, $270^\circ \pm \theta$, $360^\circ \pm \theta$, where θ is any angle.
4. Trigonometrical ratios (sine, cosine, tangent and cotangent) for the sum & difference of two angles.
5. Expressing products of sines and cosines into sum and difference form.
6. Trigonometrical ratios of multiple & sub multiple angles.
7. Angle of elevation and depression. Simple problems on height & distance.

E. GEOMETRY:

1. Pythagoras Theorem and its extension: Problems based on-
 - a. The square on one side of a triangle is greater, equal or smaller than the sum of the squares on two sides of it, according as the angle between these two sides is an obtuse angle, a right angle or an acute angle. In case of an un-equality the difference is twice the rectangle contained by one of these sides and projection of the other on this side. Knowledge of the Pythagoras number (3, 4, 5), (5, 12, 13) etc.
 - b. Apollonius Theorem- The sum of the squares contained on the two sides of a triangle equals twice the sum of the squares contained on half of the third side and the median bisecting this side.
2. Circles: Problems based on-
 - a. The line joining the center a circle to the mid point of a chord is at right angles to it.
 - b. The converse of the above theorem.
 - c. One and only one circle can pass through three points not on the same straight line.
 - d. Segments subtending equal angles at the centers of equal circles are also equal.
 - e. The converse of the above theorem.
 - f. Equal chords of equal circles are contained on equal area.
 - g. The converse of the above theorem.
 - h. In a circle equal chords are equidistant from the center.
 - i. The converse of the above theorem.
 - j. The angle subtended at the center of a circle by an area is as twice the angle subtended by it at the remaining circumference.
 - k. Angles on the same segment of a circle are equal.
 - l. If the angles subtended by a straight line formed by joining two points at two other points (in the same side) are equal, the four points are noncyclic.
 - m. Angle on the semicircle is a right angle, that on the major segment is acute and the one on the minor segment is obtuse.
3. Area and Perimeter of Plane Figures.

Official Source: IERT

 **UPDATED सिलेबस यहाँ देखें** 



IERT PRAYAGRAJ SYLLABUS



F. MENSURATION (Solid Geometry):

Surfaces & Volumes of Cube, Cuboids, Rectangular Parallelepiped, Right Prism, Right Pyramid, Right Circular Cylinder, Right Circular Cone and Sphere (application of the formulae without proof).

G. COORDINATES GEOMETRY:

Distance Between Two Points, Section Formulae, Coordinates of Centroid, Circumcentre, Incentre & Orthocentre of a Triangle, Area of a Triangle & Quadrilateral, Collinear Points, Equations of Straight Lines in Different Forms, Length of a Perpendicular, Distance Between Two Parallel Lines, Equations of Parallel & Perpendicular Lines, Coordinates of the Foot of Perpendicular, Angles Between Two Lines, Co-ordinates of a Point of Intersection, Equations of the Bisectors of the Angles Between Two Given Lines, Concurrent Lines.

PHYSICS

A. MEASUREMENTS

1. Fundamental units: Fundamental quantities, S.I. system of fundamental units, standard meter, standard kilograms, standard second, micron, Angstrom & light years, derived units.
2. Measurements of small length: Principal of vernier calipers & screw gauge and its applications, limit of accuracy of measurement, significant figures.

B. KINEMATICS:

1. Elementary Knowledge of Vectors:
Distance & displacement, speed & velocity, representation of vector, addition of vectors (triangle & polygon method-only force diagram).
2. Acceleration
Equation of motion under gravity (straight line).

C. MECHANICS:

1. Laws of Motion – Newton's law of motion, first law, second law of motion (force mass acceleration), unit of force (Newton & Kilogram weight), law of conservation of momentum, third law of motion.
2. Moment of Force – Moment of force and its unit, principal of moments, equilibrium of a body under the system of parallel forces, center of gravity, condition of stable, unstable and neutral equilibrium.

D. WORK, POWER & ENERGY:

Work, Power, Energy and its units (Joule, Watt), definition of energy, kinetic energy, potential energy (examples of elastic, gravitational and electrical energy), Transformation of energy and its relation with work, conservation of mechanical energy, different forms of energy (including mass and energy relation).

E. HYDRO-STATICS:

Pressure in liquids & its dependence on density & depth, unit of pressure, laws of liquid pressure and deduction of Archimedes's principle from it, Floatation up thrust of gases, principle of simple barometer, calculation of atmospheric pressure from the height of mercury column, practical application.

F. VIBRATION:

1. Periodic motion, time period, frequency, amplitude, characteristics of simple harmonic motion (No mathematical derivation).
2. Wave Motion & Sound-Formation of wave and its mechanism of propagation, Transverse and Longitudinal waves, wave length, speed of wave, relation between frequency of infrasonic & ultrasonic, general idea about electro-magnetic waves (gamma rays, X-rays, ultrasonic rays, visible light, infrared radiation, sources of radio waves and its simple example of application).

G. HEAT KINETIC THEORY OF MATTER:

1. Kinetic model of gases (no mathematical derivation). Explanation of the pressure of gases on the basis of kinetic theory, concept of temperature, relation between the pressure, volume and temperature of gases, ideal gas equation.
2. Thermal Expansion -Coefficient of thermal expansion of solid, liquid and gases, practical application of thermal expansion, liquid thermometer. Temperature scales (Celsius, Fahrenheit, & Kelvin).
3. Thermal Energy – Concept of internal energy, units of thermal energy (joule calorie). Thermal capacity and specific heat capacity. Latent heat, calorimeter (Principle of mixture, related problems).
4. Thermal Radiation - Dependence of rate of radiation on area, temperature & nature of surface, cooling curve, emissivity and its need.

H. LIGHT:

1. Some facts about light - Mutual independence of light rays, deviation from rectilinear propagation of light, scattering of light, diffused reflection.
2. Reflection of light – Laws of reflection, ray diagram showing the formation of images from spherical mirrors (convex & concave), its size and nature, derivation of mirror formula and its use in numerical problems, linear magnification.
3. Refraction of light on plane surface – Refraction & speed of light, Snell's law, refractive index, refraction on two parallel plane surfaces refraction, deviation, dispersion of light and total reflection in prisms.
4. Refraction through thin lenses – Converging & diverging nature of lenses; first & second principal focus. Ray diagrams showing the formation of images from lenses use of $1/V - 1/U = 1/F$ and its use in numerical problems (No derivation), linear magnification power of lenses and its unit (Diopter).
5. Optical Instruments – Formation of image in human eye power of accommodation, least distance of distinct vision, introduction of lenses as remedy of myopia (Short sightedness); hyper myopia (long sightedness), Photographic camera, slide projector, simple microscope and its magnifying power and arrangement of lens in compound microscope and astronomical telescope.

Official Source: IERT

 **UPDATED सिलेबस यहाँ देखें** 



IERT PRAYAGRAJ SYLLABUS



I. ELECTROSTATICS:

Atomic model of electrical phenomenon, negative, positive charge and their electron model, difference in conductor and non conductor (free electron model), electrostatic induction and explanation of conduction in metal on the basis of free electron model.

J. ELECTRICITY:

Simple Circuits – concept of electric current and potential difference, ampere and volt, electric cell, electromotive force, Leclanche cell, Daniell cell and drycell (with chemical reactions), polarization lead accumulator (chemical reaction not required) measurement of current and potential difference, Ohm's law, resistance & dependence of resistance on length of conductor, cross sectional area and nature of conductor specific resistance, combination of resistance in series and parallel, application of equivalent resistance in numerical problems internal resistance in numerical problems related with it. Series and parallel combination of cells.

21

CHEMISTRY

A. GENERAL & PHYSICAL CHEMISTRY

1. Science & scientific method, definition of chemistry, branches of chemistry and importance of chemistry in the modern age.
2. Matter and its three physical states & deviation based on molecular kinetic theory.
3. Atom, molecule and laws of chemical combination (Statement and examples).
4. Atomic structure, nuclear proton, neutron, atomic mass unit, electrons and their orbits (Atomic no. & isotopes) Introduction of radio activity and nuclear energy.
5. Valency- electrovalency and co-valency.
6. Symbol, formula, chemical equation, and balancing of simple chemical equations.
7. Gas laws – Boyle's laws, Charles's law, and gas equation; Gay-Lussac's law of gaseous volumes, Dalton's law of partial pressure and Graham's law of diffusion.
8. Avogadro's Hypothesis and its importance.
9. Atomic weight, molecular weight & equivalent weight. Examples and methods of their determinations (for molecular weight only Victor Meyer's and Hoffman's method).
10. Faraday's law of electrolysis, explanation of acid, base & salt, elementary concept of ionization. Principles of simple voltaic cell.
11. Different chemical reactions-addition, substitution, dissociation, hydrolysis, double decomposition, oxidation reduction and catalytic.
12. Explanation of solution- solution, solute, solvent saturated solution, unsaturated solution, and super saturated solution, normal solution-molar solution, molar solution, solubility, viz; reaction of solution, some important solvents water, benzene, ether, acetone, carbon tetrachloride and Alcohol.

B. INORGANIC CHEMISTRY:

1. Classification of elements, general characteristics of Mendeleev's periodic table.
2. Electronic configuration & position of the following elements in the periodic table, hydrogen, carbon, nitrogen, phosphorus, sulphur & chlorine.
3. Drinking water, volumetric composition of water, hard & soft water, temporary & permanent hard water. Physical & chemical methods for removal of hardness of water.
4. Laboratory method of preparation; properties and uses of the following gases; oxygen, nitrogen, chlorine, ammonia, sulphur dioxide, sulphurated hydrogen & hydrogen chloride.
5. Difference between metals & non metals, metallurgy of Aluminum.

C. ORGANIC CHEMISTRY:

1. Introduction of organic chemistry. Definition and scope of organic chemistry, difference between organic & inorganic compounds. Preliminary introduction of tetrahedral nature of carbon atom.
2. Classification & nomenclature of organic compounds-Aliphatic & Aromatic (Homologous Series).
3. Methods of preparation and properties of Paraffins (Methane and Ethane) and unsaturated hydrocarbons (Ethylene and acetylene).
4. Simple introduction of Plastic, artificial fiber, medicines, explosives, soap and detergents.

D. NUMERICAL PROBLEMS:

Numerical problems based on the following topics – laws of chemical combination, percentage composition, Empirical and molecular formula, chemical equation, equivalent weight, valency, molecular weight of Volatile substances, Graham's Law of diffusion and Gas equation.

Bharatgeek

Official Source: IERT

 **UPDATED सिलेबस यहाँ देखें** 



IERT PRAYAGRAJ SYLLABUS

